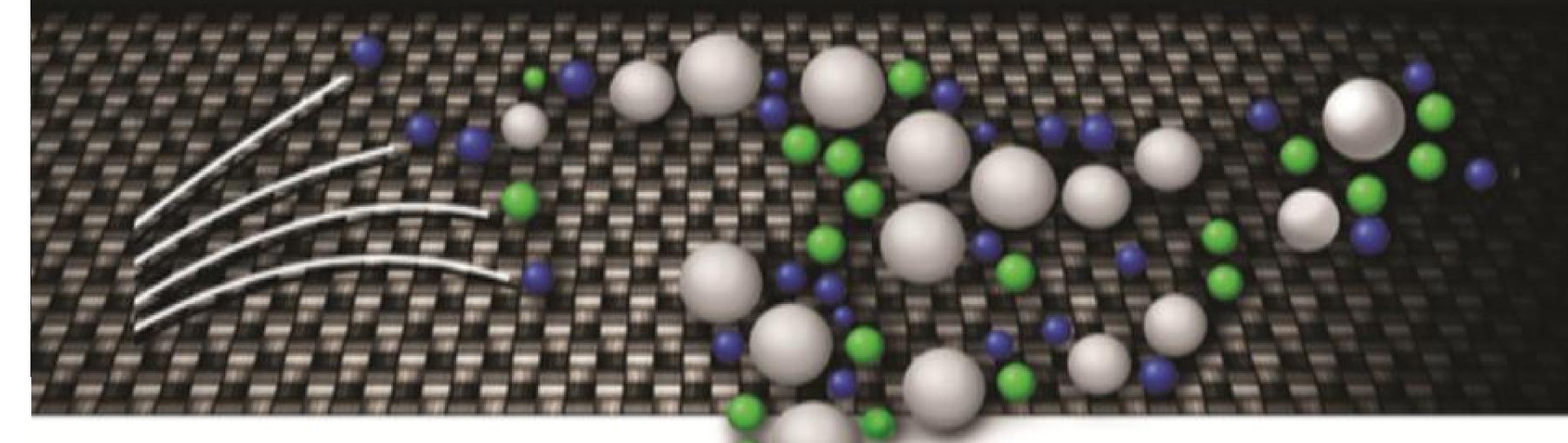
A multifunctional PvT device for the characterization of thermophysical properties of thermoplastics in extreme thermal conditions



B. Pignon⁽¹⁾, X. Tardif⁽²⁾, N. Boyard⁽¹⁾, V. Sobotka⁽¹⁾, D. Delaunay⁽¹⁾

 (1) Laboratoire de Thermocinétique de Nantes, CNRS UMR 6607, Nantes France
 (2) IRT Jules Verne, Bouguenais France baptiste.pignon@univ-nantes.fr



Introduction

"Mechanical properties of thermoplastic composites directly depend on microstructure of the matrix, which is strongly linked to the thermal history of the forming process. For modelling purpose an accurate knowledge of the thermo-physical properties and crystallization kinetics of the thermoplastic polymer is required.

The experimental determination of these parameters involves the use of many instruments, which is time consuming. To address this issue, a home-built PvT instrumented mould, dedicated to thermoplastics (including high-performance ones), has been designed to measure and/or identify several properties from a single experiment and in thermal conditions close to the process ones.

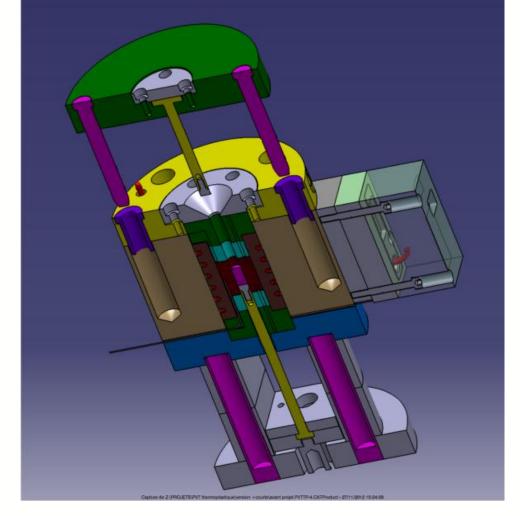
Device design and instrumentation

Thermal conductivity identification

The polymer sample is placed between two pistons with **pressure up to 200MPa**.

["]The device is heated by an induction system to temperature up to 400°C and allows high cooling rate (100K/min).

"The moulding cavity is instrumented by a heat flux sensor composed of three 40 m diameter thermocouples located at 0.25, 2.5 and 6mm from the moulding cavity wall.



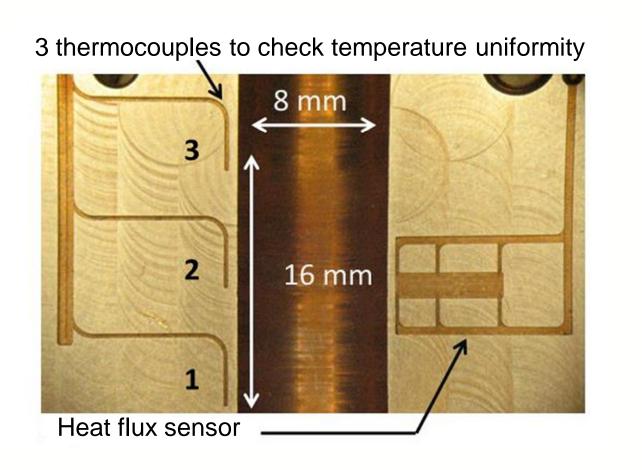


Figure 1: Sketch of the PvT-xT device

Figure 2: Instrumentation of the moulding cavity

Methodology and results : specific volume

^{*⁷*} Material : isotactic polypropylene Eltex PP HV252 (MFI ¹/₄ 11 g/10 min for 2.16 kg at 230°C) from Solvay

"A baseline is done with a ceramics standard cylinder to remove the impact of

Thermal conductivity (T), as shown in Figure 5, can be estimated in melted state by inverse method with a minimization criterion based on volumes. This method has the advantage to be **non-invasive**.

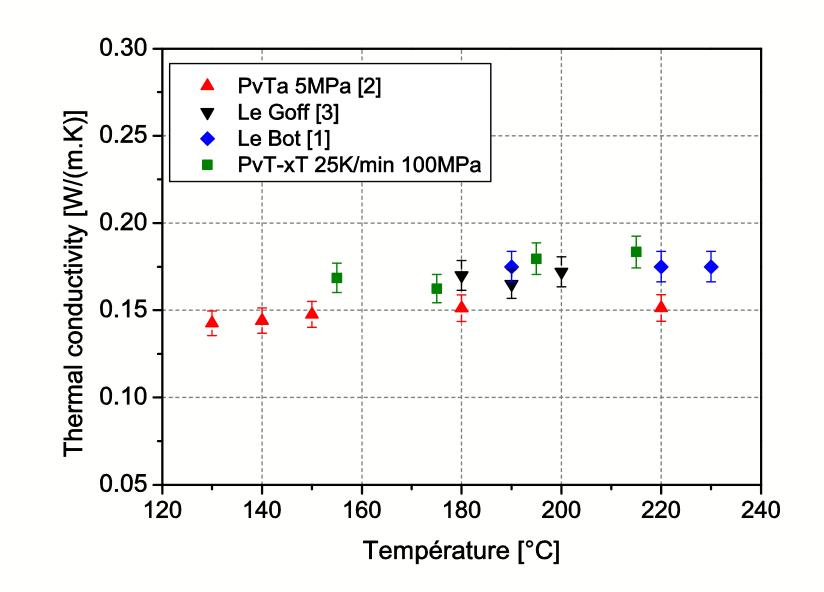


Figure 5: Identified thermal conductivity of iPP at 50K/min 60MPa compared to previous results [1-3]

Crystallization kinetics identification

Crystallization kinetics function $K_{nak}(T)$, as shown in Figure 6, can be estimated by inverse method following the same approach.



the mould thermal expansion.

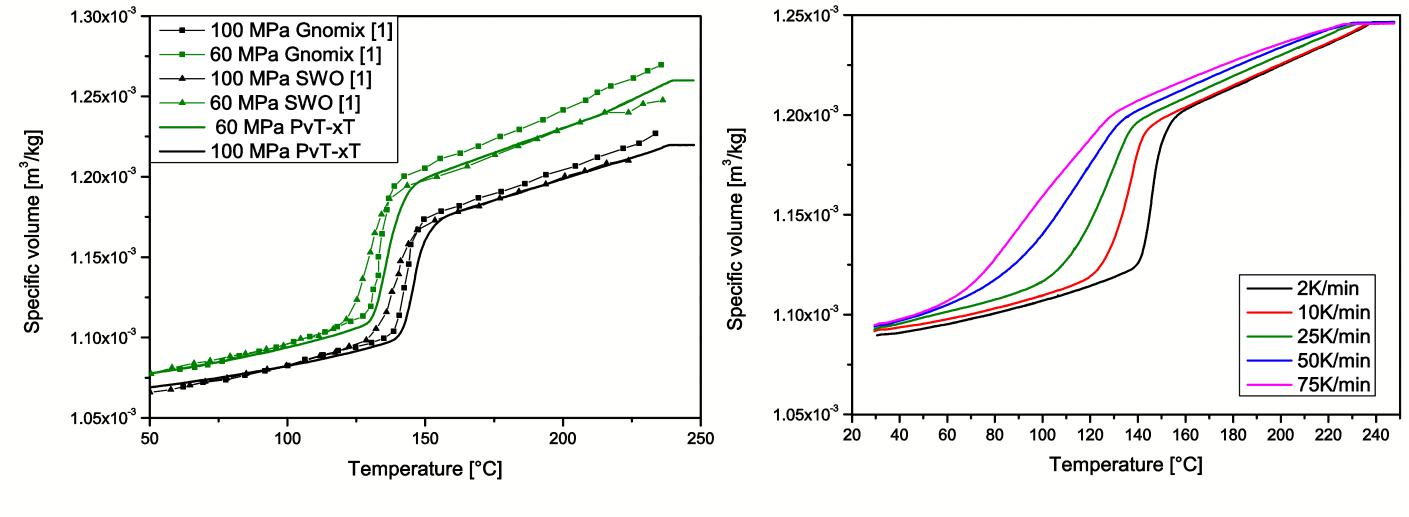


Figure 3: *iPP* specific volume 2K/min, compared to Le Bot [1]

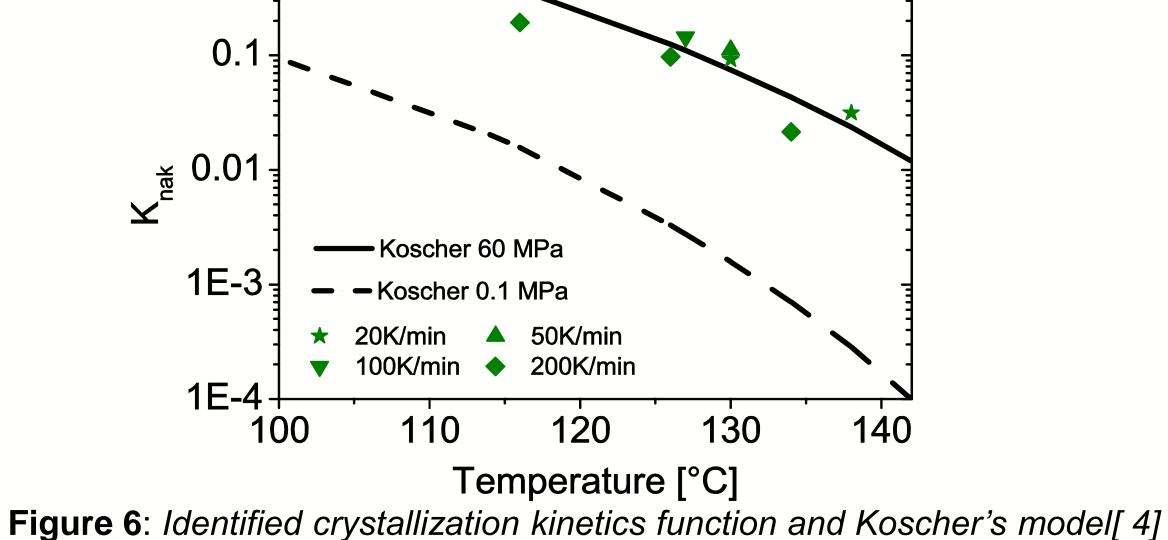
Figure 4: *iPP* specific volume for different cooling rate at 60MPa

Non-linear Heat transfer modelling

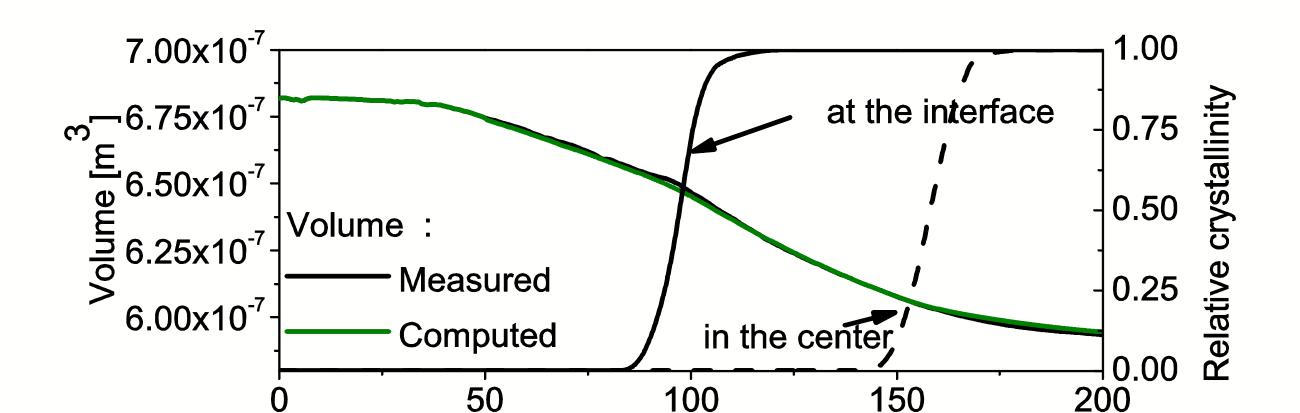
Heat transfer in the sample is modelled using the heat conduction equation, taking into account volume variations, coupled to a crystallization kinetic model.

The spatial domain is discretized in a 2D axisymmetric scheme with moving boundaries. Heat transfer is only 1D through the radius *r*.

$$\rho(P,T,\alpha).C_p(T,\alpha).\frac{\partial T}{\partial t} = \frac{1}{r}.\frac{\partial}{\partial r}\left(\lambda(T,\alpha).r.\frac{\partial T}{\partial r}\right) + \rho(P,T,\alpha).\Delta H.\frac{\partial \alpha}{\partial t}$$



After a complete identification of parameters, measured and computed volumes are well superimposed as presented in Figure 7. It highlights the strong relative crystallinity gradient through the radius of the sample at high cooling rates.



Crystallization (exothermic phenomena) is modelled by Nakamura equation.

 $\frac{\partial \alpha}{\partial t} = n.K_{Nak}(T,P).(1-\alpha).\left[-\ln(1-\alpha)\right]^{\frac{n-1}{n}}$

Figure 7: Evolution of volume and relative crystallinity for 100 K/min at 60 MPa

Time [s]

Conclusion and further research

⁷ A new home-made PvT-xT device dedicated to high performance thermoplastic polymers is presented and validated. It can reach temperature up to 400°C, pressure up to 200 MPa and cooling rate up to 100K/min

"Current works: the characterization of high performance polymers such as Polyamide (PA) and PolyEtherEtherKetone (PEEK).



[1] P. Le Bot, "Comportement thermique des semi-cristallins injectés. Application à la prédiction des retraits", Université de Nantes, 1998
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 [3] R. Le Goff, "Etude et modélisation des transferts thermiques lors de la solidification des pièces injectées en polymère semi-critsallin chargé de fibres, Université de Nantes, 2006

[4] E. Koscher, "Effets du cisaillement sur la cristallisation du polypropylène : aspects cinétiques et morphologiques," Université Claude Bernard Lyon 1, 2002.

